## UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS GCE Ordinary Level

## MARK SCHEME for the May/June 2008 question paper

## **5129 COMBINED SCIENCE**

5129/02

Paper 2 (Theory), maximum raw mark 100

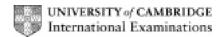
This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began.

All Examiners are instructed that alternative correct answers and unexpected approaches in candidates' scripts must be given marks that fairly reflect the relevant knowledge and skills demonstrated.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

• CIE will not enter into discussions or correspondence in connection with these mark schemes.

CIE is publishing the mark schemes for the May/June 2008 question papers for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.



Page 2			Mark Scheme	Syllabus	Paper
			GCE O LEVEL – May/June 2008	5129	02
1	(a)	(i) tree	/ grass / flower		[1]
		(ii) cow	/ snail / rabbit		[1]
		Box 2 = 9 Box 4 = 1			[2]
	(c)	Sun / sur	nlight (not light alone)		[1]
	(d)	decompo	oser / bacteria / fungi		[1]
		energy is	upply is limited s used by the organisms any 2 s lost at each stage		
			nt energy left (for another level)		[2]
2	(a)	copper /	Cu		[1]
	(b)	potassiu	m / K		[1]
	(c)	iron / Fe			[1]
	(d)	copper /	Cu		[1]
	(e)	zinc / Zn			[1]
3	(a)	` '	It or 0.2 × 180 0.6 gains 1 mark)		[2]
		(ii) V = 1 = 1.4	IR or 7 × 0.2 4		[2]
	(b)	0.6 / 2.0-	-(a)(ii)		[1]

Page 3	Mark Scheme	Syllabus	Paper	
	GCE O LEVEL – May/June 2008	5129	02	

**4** (a) limewater milky / cloudy / white (precipitate)

[2]

(b) (i) 
$$CH_4 = 16$$
  
 $CO_2 = 44$ 

[2]

(ii) 
$$16 \rightarrow 44$$
  
 $\therefore 4 \rightarrow 44 \times 4/16 = 11 \text{ g}$   
correct method from wrong numbers in **(b)(ii)** gains 2

[2]

5 (a) blue pink

[1]

(b) (i) transpiration

[1]

(ii) upper surface has waxy layer fewer / no stomata answer could be in terms of lower surface

[2]

(c) root hair osmosis

[2]

6 (a) reduction

[1]

(b) conducts electricity conducts heat malleable ductile high density high melting point

high boiling point

(c) boils at 100 °C/boils at single temperature

[1]

[2]

7 (a) (i) gravity / weight

[1]

(ii) gravitational / potential

[1]

(b) line is curved / not straight

[1]

(c) F = ma or a = F/m or 300/80= 3.75 $m/s^2$ 

[3]

any 2

	Pa	ge 4	Mark Scheme	Syllabus	Paper					
		_	GCE O LEVEL – May/June 2008	5129	02					
8	(a)	a) matt black is a better absorber / shiny is a better reflector								
	(b)	(i)	stays the same / no change / none		[1]					
		(ii)	decreases / gets less / lowers		[1]					
	(c)	micr	owaves and radiowaves (either order)		[2]					
9	(a)	large mixe	/ grinds food e pieces to smaller pieces es food with saliva plyes (soluble particles)	any 2	[2]					
	(b)	secr lubri	ete liquid / saliva ete enzymes / amylase cate / softens food rmes convert starch to maltose / sugar	any 2	[2]					
	(c)	bact cavit	eria ies / enamel to dissolve / tooth decay		[2]					
10	(a)	hydr	ogen / H⁺		[1]					
	(b)	(i)	red		[1]					
		(ii)	orange / yellow		[1]					
	(c)	(i)	$Mg + H_2SO_4 \rightarrow MgSO_4 + H_2$		[1]					
		` ,	magnesium carbonate magnesium hydroxide any 2 magnesium oxide		[2]					
11	(a)	like (	charges (repel)		[1]					
	(b)	posi	ive		[1]					

	. u	900	GCE O LEVEL	- May/June 2	2008	5129	02
12		uency = I od = s	Hz or s <sup>-1</sup>				[2]
13	(a)	and one	showing 3 bonding pairs lone pair shell drawn it must be cori	rect)			[2]
	(b)	400–500 200–300 iron					[3]
	(c)		m rus (either order)				[2]
14	(a)	lack of (e	enough) food				[1]
	(b)	not enou too much civil unre earthqua	rain / water / floods	ırks	plants die / cr no photosynti plants washe no one to ten or crops dest food destroye	nesis / growth d away / die d crops royed	[4]
15	(a)	both pos	al shape with itive and negative values cycles shown				[3]
	(b)	increase stronger more turi			any 1		[1]
16	(a)	0.1 × 30 W = 0.1	= W × 0.2 5				[2]
	(b)		lockwise / iron rod goes dattracted by the magnet	lown / left goe	s down		[2]

Mark Scheme

Syllabus

Paper

Page 6							Mark Scl	heme			Syllabus	Paper
					G	CE O L	EVEL – N	May/June 200	08		5129	02
17	(a)	) (i) electronic structure drawn as 2 8								[1]		
		(ii)	+3									[1]
	(b)	on th	is a p ne let	ft of t		odic Ta s metal	ble to non-m	etallic		any	2	[2]
	(c)						iinium) ox of metal	kide				[2]
18	(a)	A = 1 B = 0 C = 1 D = 1	cotyl plum	ledon nule								[4]
	(b)	wate oxyg suita	jen	/ nam	ed tem	peratur	е					[3]
19	(a)	26 – (one				from dia	agrams ga	ains 1 mark)				[2]
	(b)	0.24	or	<b>(a)</b> /5	0							[1]
20	(a)	) 5							[1]			
	(b)	extension = 10 (cm) load = 4								[2]		